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The Impact of Burner Staging on NO Reburning during Oxy-coal Combustion

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Oxy-coal Combustion: An Introduction

- Oxy-fuel combustion is a carbon capture technology
- An O_2/CO_2 oxidant is used instead of air in order to produce a flue gas with a far higher CO_2 content
- Greatly simplifying CO_2 capture
- The oxidant is formed by recirculating flue gas and combining with pure O_2
- Oxy-fuel combustion has been called the most techno-economically feasible CCS technology



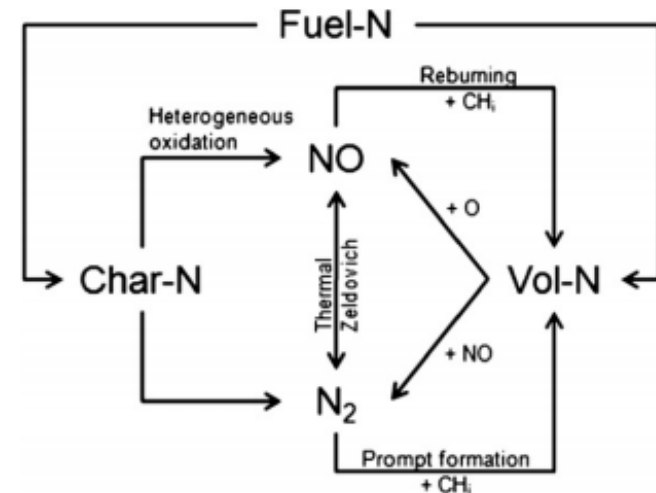
Oxy-coal Combustion: Common Challenges

- Oxidant has some very different properties to air, including a higher heat capacity and density
- This impacts flame temperature and stability
- In order to match air's flame temperature the oxygen concentration in the oxidant must be enriched
- The technology is associated with costly unit operations (ASU, SCR etc)



Oxy-coal Combustion: NO_x Processes

- NO concentration in flue gas tends to be higher than air
- Due to lack of nitrogen's diluting effect
- Emission rate is lower though
- Higher O_2 concentration enabling increased conversion of fuel-N \rightarrow NO
- Recycled NO reburning \rightarrow reduction of NO through reaction with volatile-C, volatile-N and char (in presence of CO)

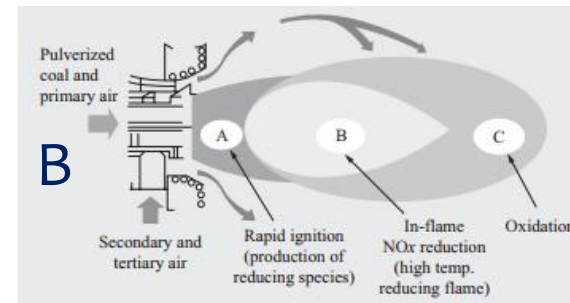
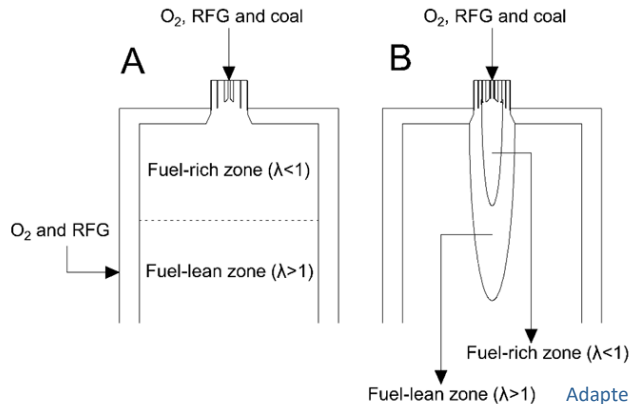


Toftegaard, M.B., Brix, J., Jensen, P.A., Glarborg, P., Jensen, A.D. (2010) Oxy-fuel combustion of solid fuels. *Progress in Energy and Combustion Science*, 36(5), 581-625.



Oxy-coal Combustion: Burner Staging

- Commonly, studies utilise furnace staging (A) in order to minimise NO formation
- However, the presence of NO in an over-fire stream could reduce overall rate of NO reburning
- Studying the impact of burner staging (B) on NO reburning is needed
- Increased in-flame NO reduction would reduce load on secondary NO_x technologies and help realise zero-NO_x oxy-coal combustion



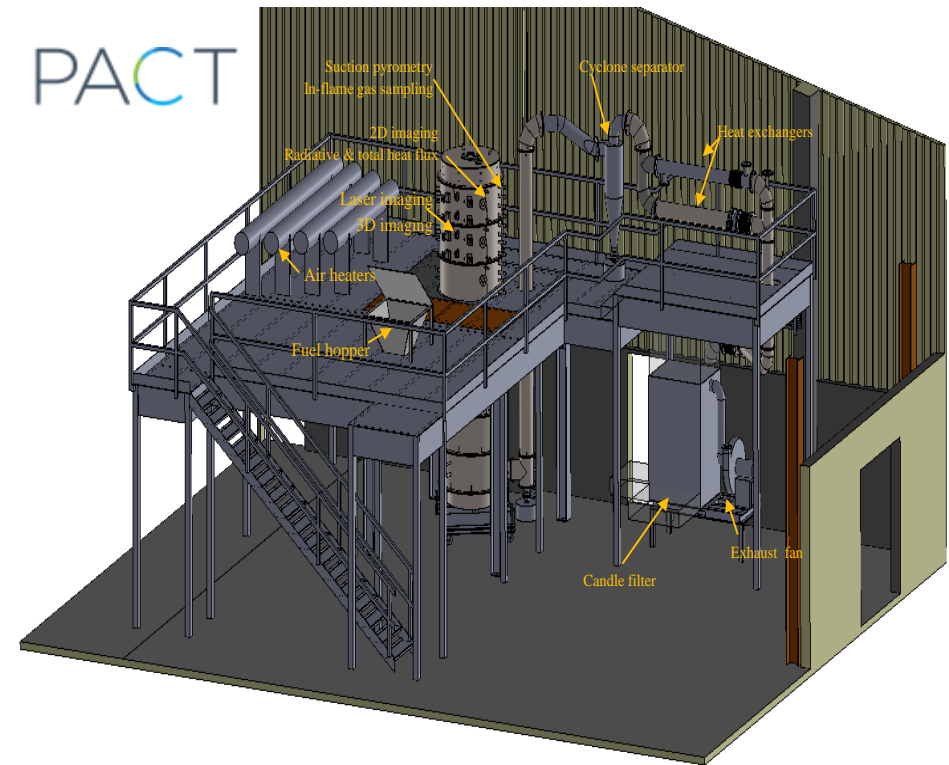
Adapted from: Ochi, K., Kiyama, K., Yoshizako, H., Okazaki, H., Taniguchi, M. (2009) Latest low-NO_x combustion technology for pulverised-coal-fired boilers. Hitachi Review, 58(5), 187-193.

Adapted from: Normann, F., Andersson, K., Leckner, B., Johnsson, F. (2009) Emission control of nitrogen oxides in the oxy-fuel process. Progress in Energy and Combustion Science, 35, 385-397.



Experimental Setup

- 250 kW_{th} combustion test facility at PACT in Sheffield, UK
- Conditions: Air and OF 28 at 200 kW_{th} and OF 27 and OF 30 at 170 kW_{th}
- Measurements taken radially, axially and in the flue
- Oxidant with recycled flue gas is simulated using pure CO₂, O₂ and NO



Szuhanszki, J., Farias Moguel, O., Finney, K., Akram, M., Pourkashanian, M. (2017) Biomass combustion under oxy-fuel and post combustion capture conditions at the PACT 250 kW air/oxy-fuel CTF. Available: http://www.supergen-bioenergy.net/media/eps/super-gen/presentations/assembly-2017/25.10.2017_SUPERGEN---Sheffield-Project-outputs_for-web.pdf



Operation of the Burner

- Initial (1°) oxidant and burnout (combined 2° and 3°) oxidant mass flows are controlled
- Sliding damper on the burner allows partitioning of the burnout oxidant into variable 2° and 3° flows, while 1° remains constant
- This enables variability of stoichiometry in the fuel-rich region and overall swirl of the flame
- The secondary oxidant proportion (S): ratio of mass flow of oxygen in the 1° and 2° oxidant to the mass flow of oxygen in the combined 2° and 3° oxidant

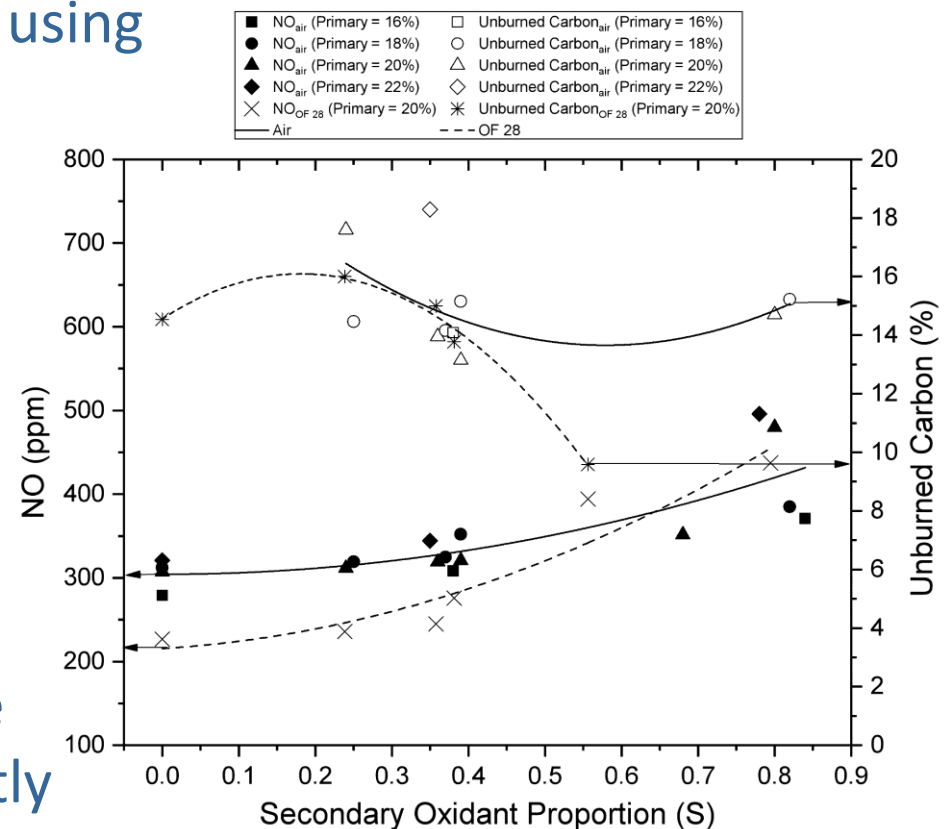


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Impact of the Primary Flow and Burner Staging on NO Formation

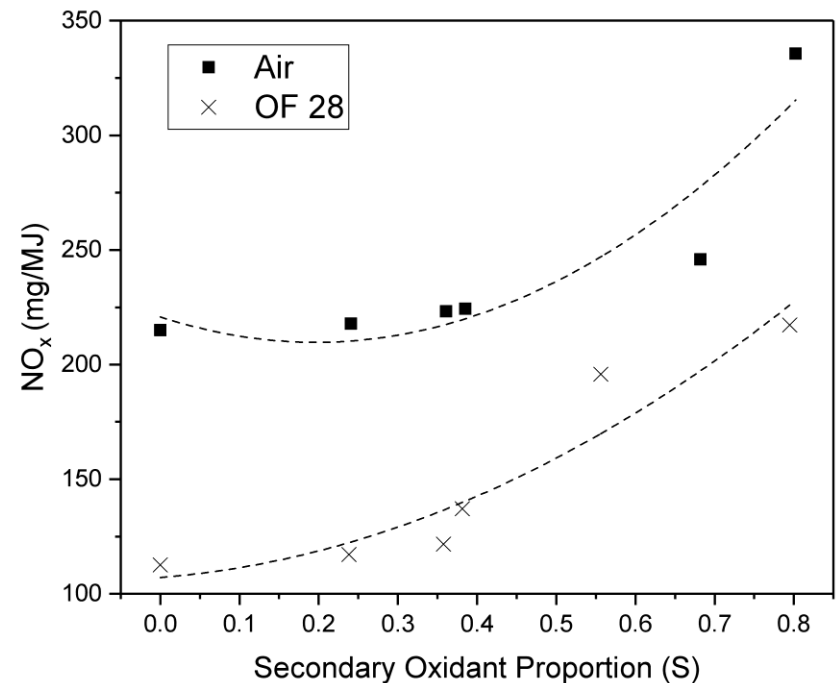
- Four 1° oxidant flowrates tested using air (16%, 18%, 20%, 22%)
- 20% most favourable setting across range of secondary oxidant proportions
- This is now used as a constant for all oxy-coal scenarios
- At OF 28, NO emissions are lower than air until $S < 0.5$, while unburned carbon only significantly improves when $S > 0.5$





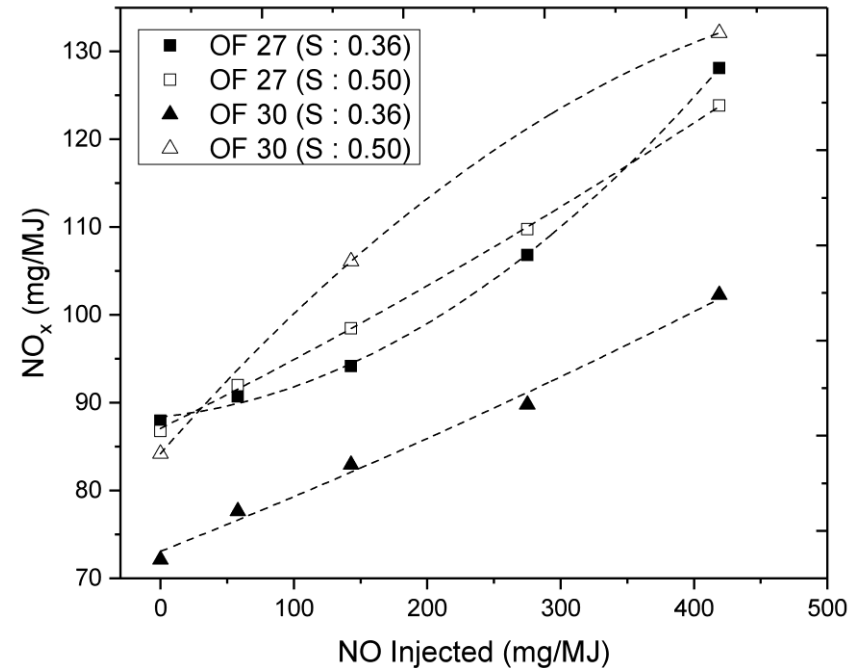
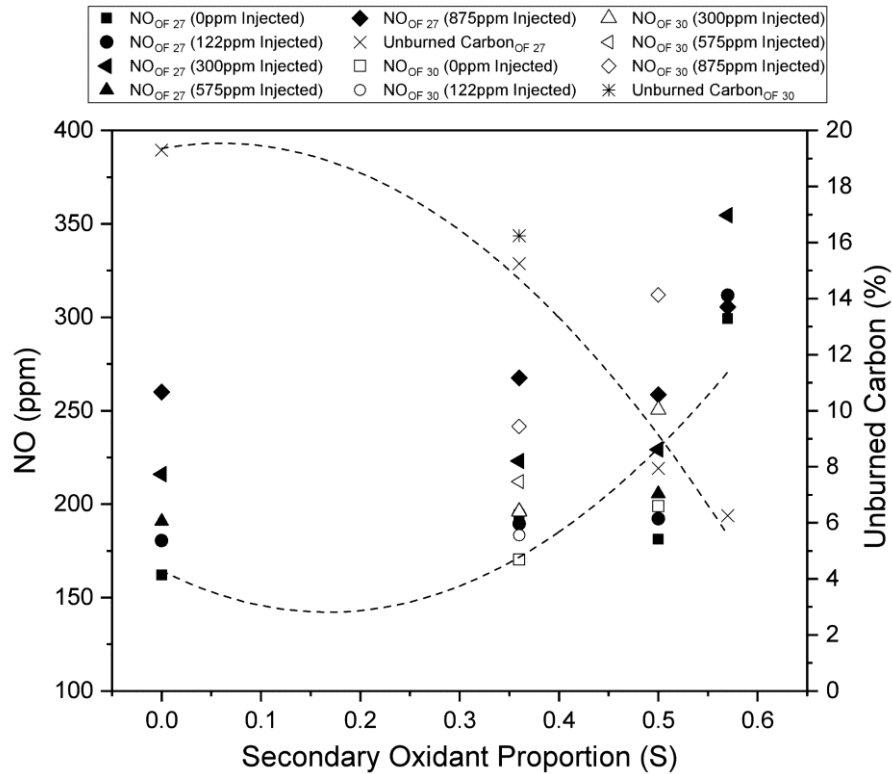
Comparison of NO_x Emission Rate without NO 'Recycling'

- Confirmation of far superior emission rate from a staged oxy-coal flame
- Reasons for lower NO_x formation:
 - Lack of thermal and prompt NO
 - The reverse Zeldovich mechanism
 - Likely temperature increase in the fuel-rich zone causing:
 - reduced char-N → NO,
 - increased fuel-N → volatile-N,
 - increased volatile-N → N₂



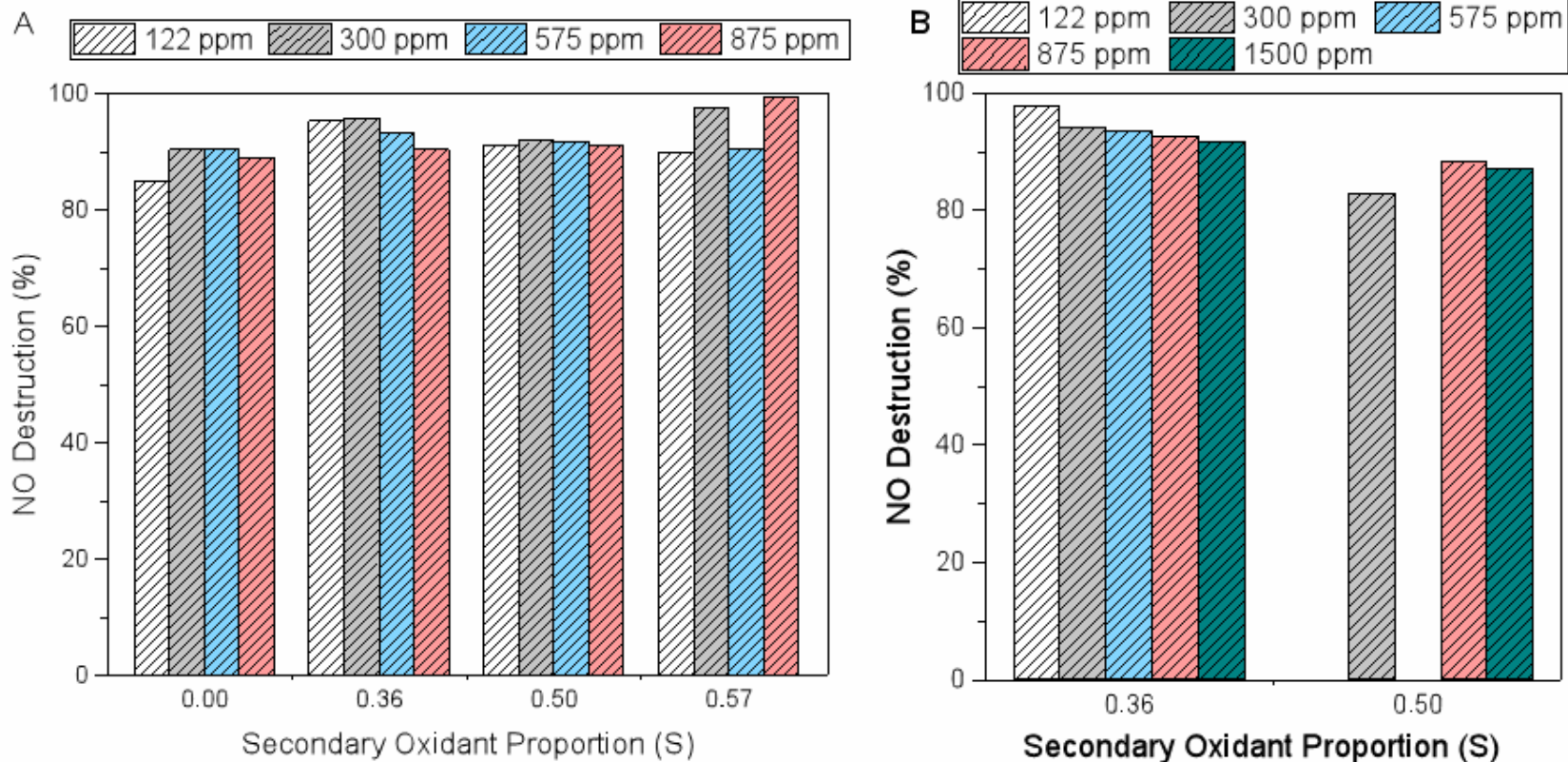


Impact of NO 'recycling' and Burner Staging on Combustion



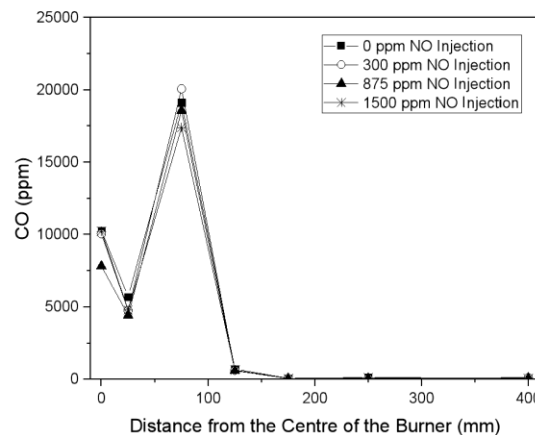
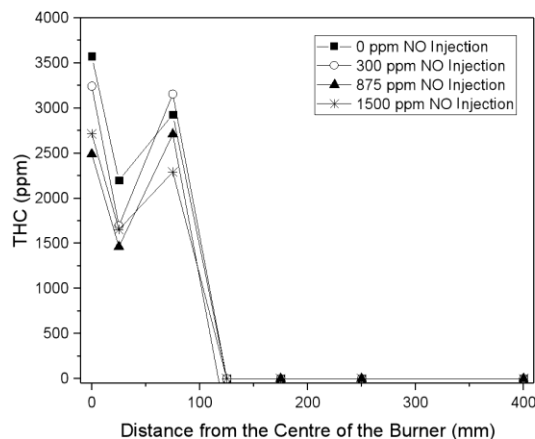
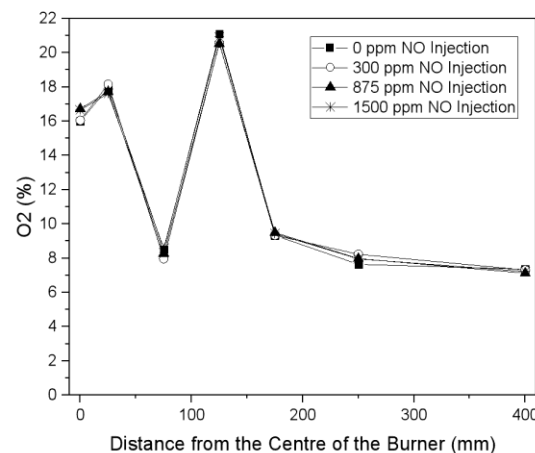
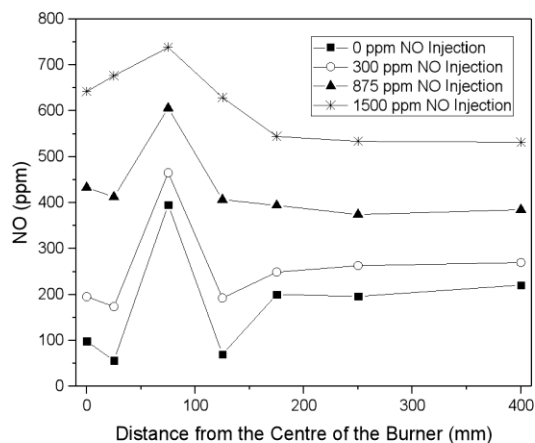


Impact of Near-Burner Stoichiometry on NO Reburning



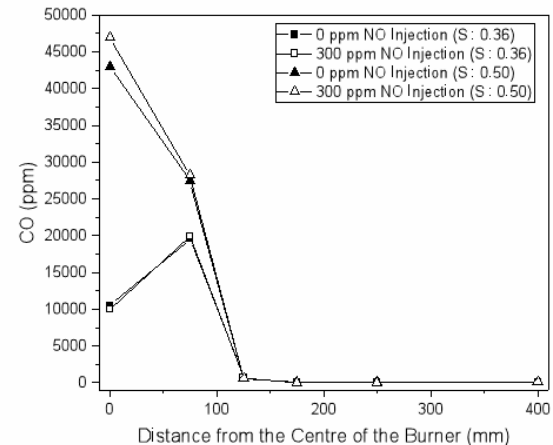
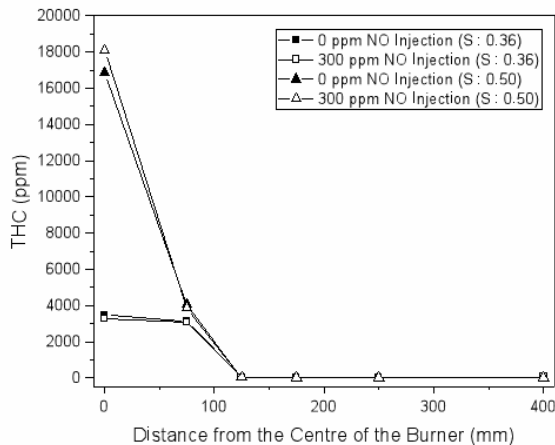
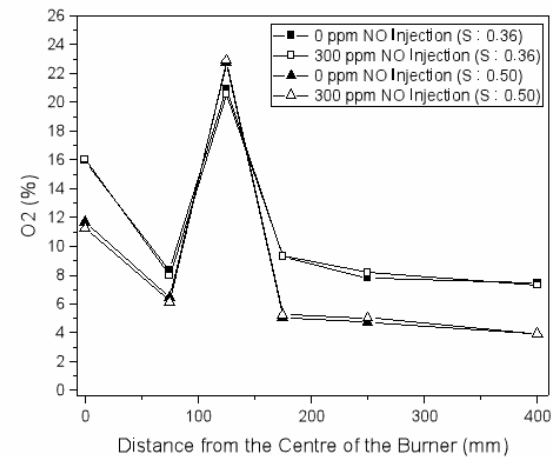
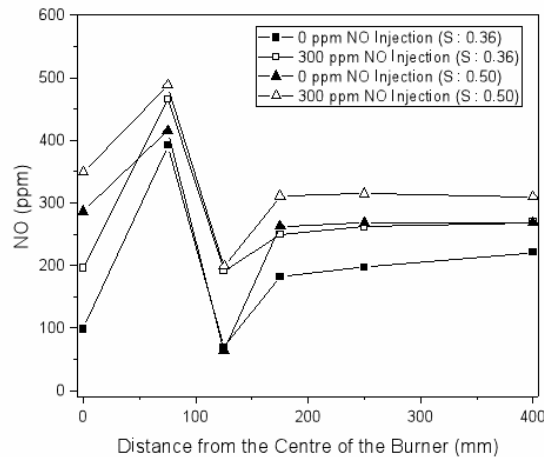


Impact of NO 'Recycling' on Radial Profile of Key Flame Constituents (S: 0.36)



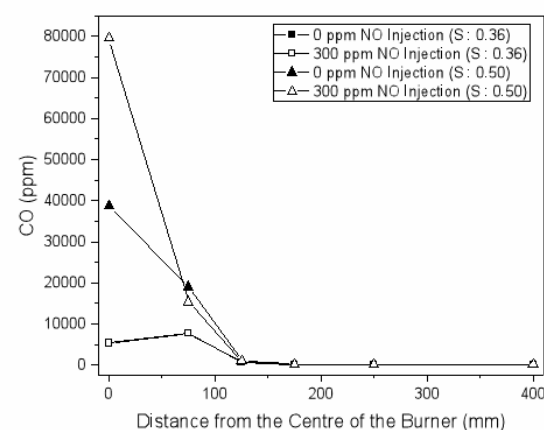
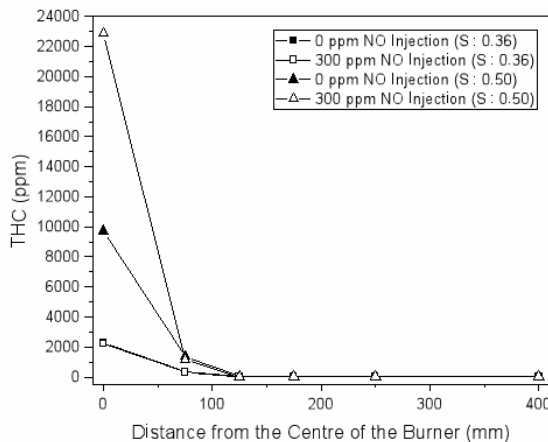
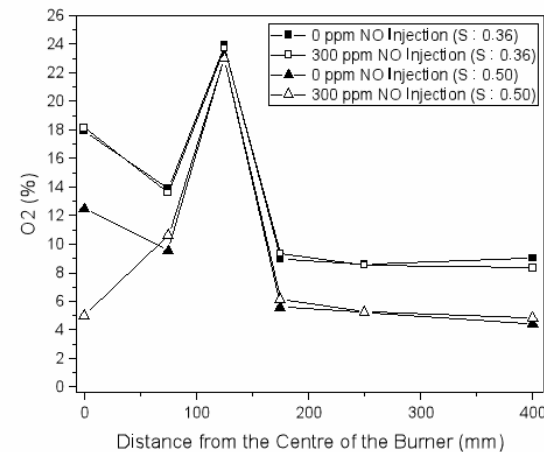
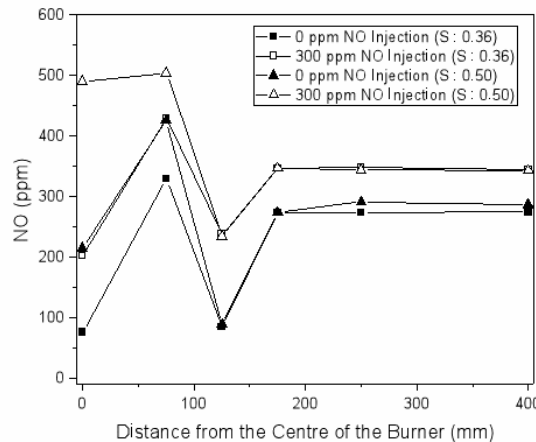


Impact of Varied Burner Staging Environments on Radial Profile of Key Flame Constituents



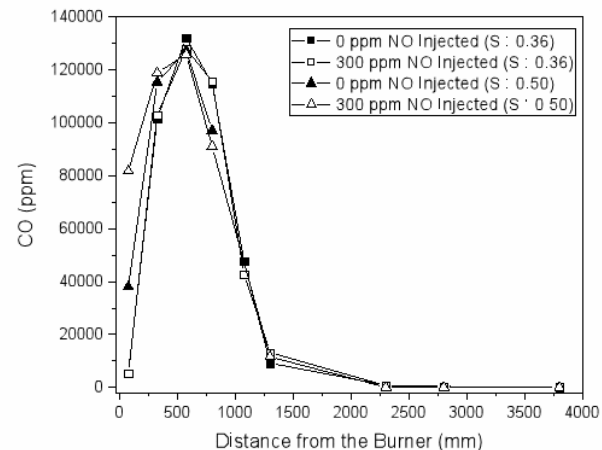
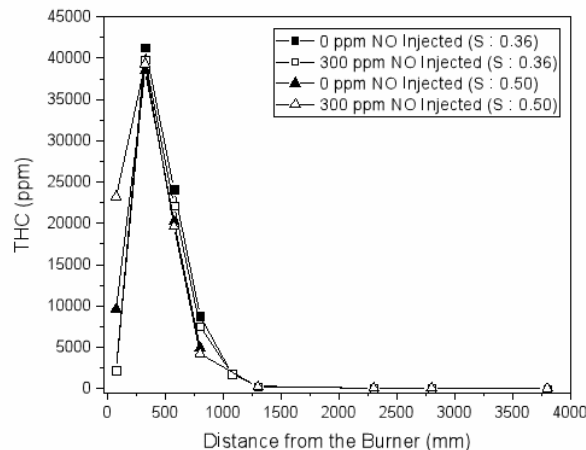
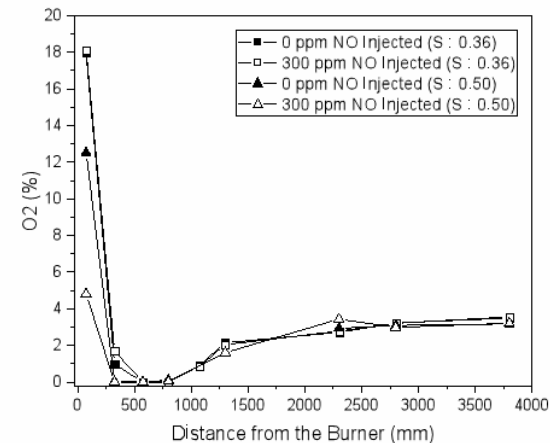
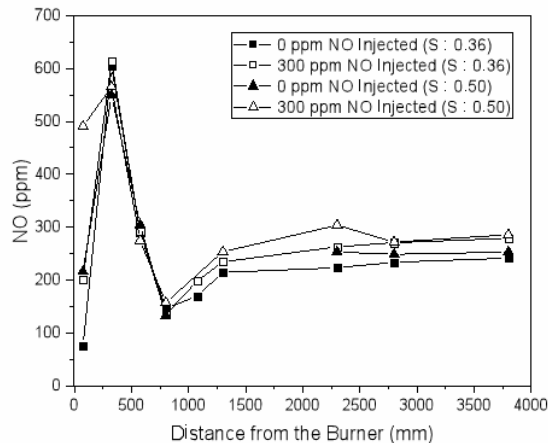


Impact of Varied Burner Staging Environments on Radial Profile of Key Flame Constituents





Impact of Varied Burner Staging Environments on Axial Profile of Key Flame Constituents



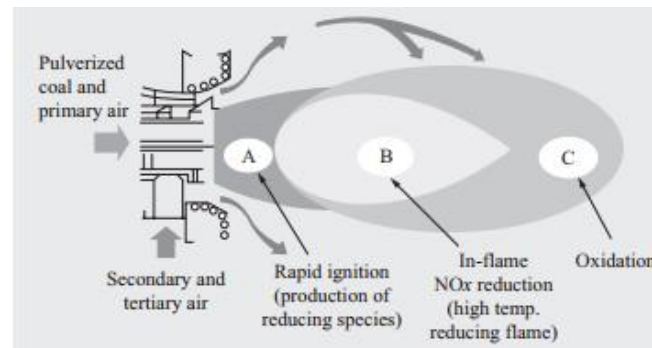
Conclusions

Comparison of burner staging impact on oxy flames and air flames:

- Prominently reduced NO_x formation
- Decreased sensitivity to burner staging

Comparison of burner staging impact on different oxy flames:

- Reduced NO_x formation at higher O_2 concentration
- Increased NO_x reduction at higher O_2 concentration
- Recycled NO is almost immediately destroyed (A), therefore control of by-products in the reducing zone (B) is very favourable



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Thank you for listening, any questions?

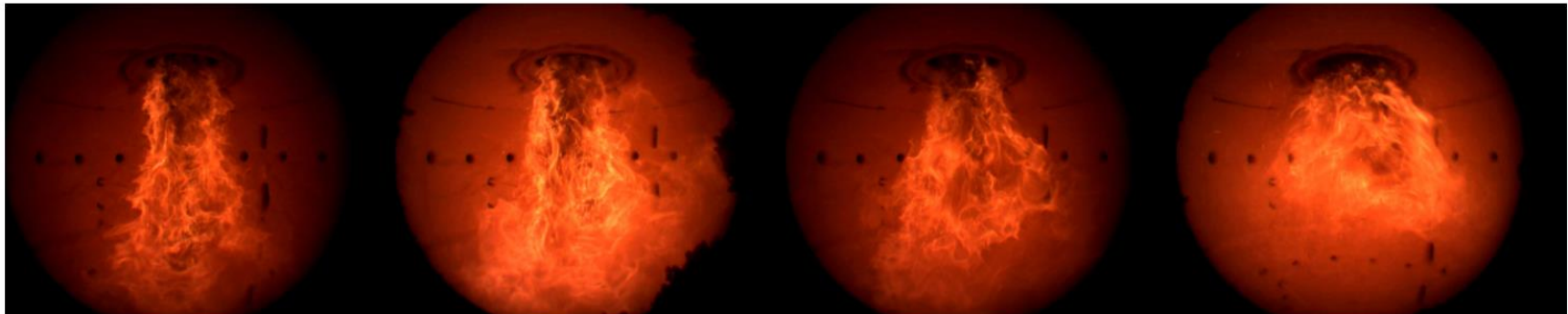
OF 27

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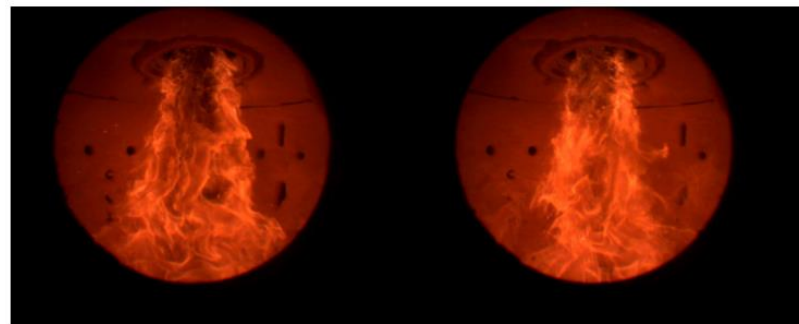
S : 0.50

S : 0.57



S : 0.36

S : 0.50



OF 30